Alkyl Monolayers on Silica Surfaces Prepared from Neat, Heated CISi(CH₃)₂(CH₂)₆CH=CH₂ Analyzed by XPS

Ghaleb A. Husseini, Amarchand Sathyapalan, Eric T. Sevy, Matthew R. Linford, and Matthew C. Asplund

Dept. of Chemistry and Biochemistry, Brigham Young University, Provo, UT

Silane monolayers on silica, prepared from mono-, di-, and trichlorosilanes, are widely used in industry for surface functionalization and modification. However, unlike di-and trichlorosilanes, monochlorosilanes are particularly easy to work with because they can dimerize, but not polymerize, upon reaction with water. Typically, an organic solvent is used when depositing a silane monolayer. Here we show XPS spectra of monolayers of ClSi(CH₃)₂(CH₂)₆CH=CH₂ (octenyldimethylchlorosilane, CAS# 17196-12-2) on silicon oxide (silicon wafer) prepared using a rapid, solvent-free approach. Reaction conditions are 120 °C for 10 min using the neat (pure) compound, and no inert atmosphere or special treatment of the compound is required. © 2003 American Vacuum Society. [DOI: 10.1116/11.20020503]

Keywords: x-ray photoelectron spectroscopy; silane; alkylation; monochlorosilanes

PACS: 82.80.Pv, 81.05.Lg

Host Material: Alkyl monolayer/Si – CISi(CH₃)₂(CH₂)₆CH=CH₂
Instrument: Surface Science SSX-100
Major Elements in Spectrum: C, Si, O
Minor Elements in Spectrum: none

Printed Spectra: 7

Accession # 00737

Technique: XPS

Spectra in Electronic Record: 21
Spectral Category: technical
Original Submission: 5/23/2002
Accepted for Publication:
9/30/2002

. .

INTRODUCTION -

Silanes (Ref. 1) attach to silica surfaces by reacting with surface silanols (Si-OH). While monolayers prepared from monochlorosilanes are more subject to hydrolysis than those derived from diand trichlorosilanes, surface functionalization with the mono-functionalized compounds is generally more straightforward because they cannot polymerize. Numerous preparations of silane monolayers on surfaces have been reported in the literature. For example, Maoz and Sagiv first showed that alkyl-terminated monolayers can be prepared on planar silicon using trichlorosilanes (Ref. 2). Recognizing the advantages of mono- over di- and trichlorosilanes, Rabolt and co-workers described a gas phase properfluorinated depositing alkyldimethylchlorosilanes on SiO₂ (Ref. 3). Both Linton and co-workers (Ref. 4) and Watts and co-workers (Ref. 5) deposited monolayers of alkyldimethylchlorosilanes, onto silica particles using an organic solvent. Typical surface modification conditions call for exposing a surface to a heated, dilute solution of a silane under inert atmosphere. In contrast, here we take clean, dry, native-oxide-terminated silicon (1.5-2.0 nm), place ClSi(CH₃)₂(CH₂)₆CH=CH₂ on its surface, and heat it at 120 °C for 10 min (relative humidity = 70%). After reacting, the surface is cleaned, dried, loaded into an XPS UHV chamber, and analyzed by XPS, which showed carbon levels consistent with monolayer quantities of surface alkyl chains [ellipsometric thickness = 0.72 ± 0.14 nm, advancing contact angle $(\theta_a) = 90.5^{\circ} \pm 0.5^{\circ}$], receding contact angle $(\theta_r) = 76.0^{\circ} \pm$ 2.2°.

Three replicate samples were subjected to identical treatment and analysis, to show the reproducibility of our technique. Only one sample, with its spectra, is published here, except that the Table of Spectral Features lists comparable features from all three samples. In addition, complete data and spectra for all three samples are archived in the Surface Science Spectra database.

SPECIMEN DESCRIPTION -

Host Material: Alkyl monolayer on clean, native oxide terminated silicon derived from ClSi(CH₃)₂(CH₂)₆CH=CH₂

CAS Registry #: 7440-2-13

Host Material Characteristics: homogeneous; solid; single crystal; semiconductor; glass; thin film; coating

Chemical Name: silicon/silicon oxide

Source: Montco Silicon Technologies, Inc.

Host Composition: Si/SiO₂

Form: single crystal wafer, p-type

Lot #: W9969 sample 4

Structure: Si(100)

As Received Condition: silicon wafer, 125 mm diameter

Analyzed Region: host material plus prepared monolayer

Ex Situ Preparation/Mounting: The silicon surfaces were first cleaned with a solution of NH₄OH (conc.): H₂O₂ (conc.) (50:50) (v/v) for 30 min at room temperature. They were then rinsed with water and finally washed with 5% vol. HCl (conc.) for 1 h. After reaction in the oven (for 10 min), the wafers were rinsed with acetone, cleaned with a soft artists brush using a 2% sodium dodecyl sulfate solution in water, and placed in a Soxhlet apparatus overnight using m-xylene (b.p. \sim 139 °C) as the extraction solvent. The samples were then removed from the Soxhlet, rinsed with water, dried, and mounted into the XPS machine. (Note: source beam size on the instrument was not well characterized and may be up to twice as large as the manufacturer's values given here [See entry for Source Beam Size at Specimen Surface].) Warning: this procedure should not be attempted with volatile silanes. Fumes from a volatile organic compound are potentially explosive. In addition, the NH₄OH/H₂O₂ cleaning solution is extremely caustic and should be used with great care.

In Situ Preparation: not specified

Charge Control: none

Temp. During Analysis: 298 K

Pressure During Analysis: $<1.79\times10^{-7}$ Pa

INSTRUMENT DESCRIPTION .

Manufacturer and Model: Surface Science Laboratories, Inc.,

SSX-100

Analyzer Type: spherical sector

Detector: resistive anode position detector **Number of Detector Elements:** 128

INSTRUMENT PARAMETERS COMMON TO ALL SPECTRA

■ Spectrometer

Analyzer Mode: constant pass energy

Throughput ($T = E^N$): N = 0

Excitation Source Window: $10 \mu m$ Mylar Excitation Source: Al K_{α} monochromatic

Source Energy: 1486.6 eV **Source Strength:** 200 W

Source Beam Size: $0.8~\mathrm{mm} \times 0.8~\mathrm{mm}$ Signal Mode: multichannel direct

■ Geometry
Incident Angle: 55°

Source to Analyzer Angle: 70.8°

Emission Angle: 55°

Specimen Azimuthal Angle: 0°

Acceptance Angle from Analyzer Axis: 30° Analyzer Angular Acceptance Width: $30^{\circ} \times 30^{\circ}$

DATA ANALYSIS METHOD

Peak Shape and Background Method: background Shirley func-

tion

ACKNOWLEDGMENTS -

The authors acknowledge the help of Yit-Yian Lua at Brigham Young University in using the XPS.

REFERENCES -

- E. P. Plueddemann, Silane Coupling Agents (Plenum, New York, 1991).
- 2. R. Maoz and J. Sagiv, J. Colloid Interface Sci. 100, 465 (1984).
- P. W. Hoffman, M. Stelzle, and J. F. Rabolt, Langmuir 13, 1877 (1997).
- S. J. Simko, M. L. Miller, and R. W. Linton, Anal. Chem. 57, 2448 (1985).
- V. A. Brown, D. A. Barrett, P. N. Shaw, M. C. Davies, H. J. Ritchie, P. Ross, A. J. Paul, and J. F. Watts, Surf. Interface Anal. 21, 263 (1994).

SPECTRAL FEATURES TABLE								
Spectrum ID #	Element/ Transition	Peak Energy (eV)	Peak Width FWHM (eV)	Peak Area (counts)	Sensitivity Factor	Concentration (at. %)	Peak Assignment	
00737-02	Si 2p	100.00	1.744	49165	0.9	42.12	•••	
00737-03	C 1s	285.70	3.127	26524	1.0	20.533	• • •	
00737-04	O 1s	533.13	1.998	120269	2.49	37.344	• • •	
00737-05	Si 2p	99.94	1.257	10839	0.9			
00737-06	C 1s	285.97	1.671	1886	1.0	• • •	• • •	
00737-07	O 1s	533.12	2.951	28488	2.49			
00737-09	Si 2p	99.98	1.423	52907	0.9	41.796		
00737-10	C 1s	285.64	2.741	27256	1.0	19.454		
00737-11	O 1s	533.17	2.489	135341	2.49	38.750		
00737-12	Si 2p	99.99	1.196	10467	0.9			
00737-13	C 1s	285.6	1.80	5591	1.0			
00737-14	O 1s	533.06	3.018	30507	2.49			
00737-16	Si 2p	101.43	4.83	28571	0.9	43.642	•••	
00737-17	C 1s	286.07	3.546	13648	1.0	18.826		
00737-18	O 1s	533.08	2.205	67850	2.49	37.532		

SPECTRAL FEATURES TABLE (CONT.)

Spectrum ID #	Element/ Transition	Peak Energy (eV)	Peak Width FWHM (eV)	Peak Area (counts)	Sensitivity Factor	Concentration (at. %)	Peak Assignment
00737-19	Si 2p	101.5	4.65	5164	0.9	•••	•••
00737-20	C 1s	285.97	2.438	2168	1.0	• • •	• • •
00737-21	O 1s	533.31	2.295	13327	2.49	•••	•••

ANALYZER CALIBRATION TABLE

Spectrum ID #	Element/ Transition	Peak Energy (eV)	Peak Width FWHM (eV)	Peak Area (counts)	Sensitivity Factor	Concentration (at. %)	Peak Assignment
a	Au $4f_{7/2}$	83.92	0.98	2200	10.67	• • •	
b	Au $4f_{7/2}$	83.92	1.6	6000	10.67	• • • •	•••
c	Cu 3s	122.36	3.0	1600	1.05	• • •	•••
b	Cu $2p_{3/2}$	932.45	1.78	4000	9.73	•••	•••

 $^{^{\}rm a}$ Spot size 300 $\mu{\rm m},$ pass energy 50 eV, 2 scans.

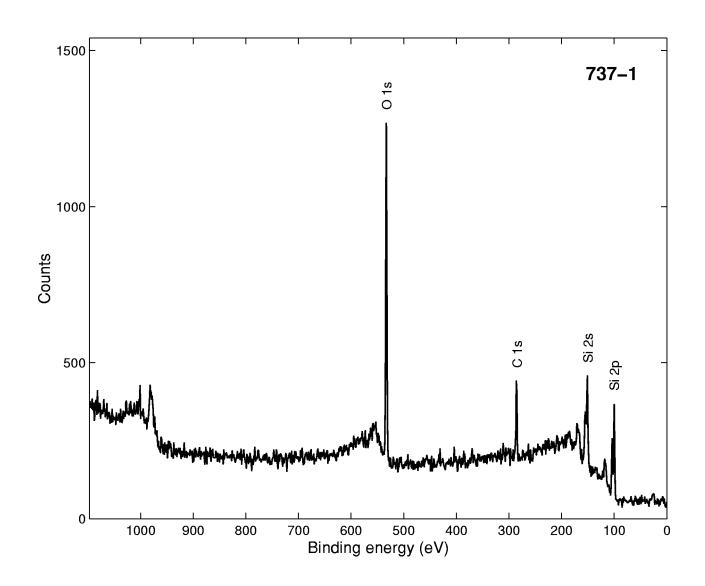
GUIDE TO FIGURES

Spectrum (Accession) #	Spectral Region	Voltage Shift*	Multiplier	Baseline	Comment #
737-1	Survey	0	1	0	
737-2	Si 2 <i>p</i>	0	1	0	
737-3	C 1s	0	1	0	
737-4	O 1s	0	1	0	
737-5	Si 2p	0	1	0	
737-6	C 1s	0	1	0	
737-7	O 1s	0	1	0	
738-1 [NP]**	Survey	0	1	0	
738-2 [NP]	Si 2p	0	1	0	
738-3 [NP]	C 1s	0	1	0	
738-4 [NP]	O 1s	0	1	0	
738-5 [NP]	Si 2p	0	1	0	
738-6 [NP]	C 1s	0	1	0	
738-7 [NP]	O 1s	0	1	0	
739-1 [NP]	Survey	0	1	0	
739-2 [NP]	Si 2p	0	1	0	
739-3 [NP]	C 1s	0	1	0	
739-4 [NP]	O 1s	0	1	0	
739-5 [NP]	Si 2p	0	1	0	
739-6 [NP]	C 1s	0	1	0	
739-7 [NP]	O 1s	0	1	0	

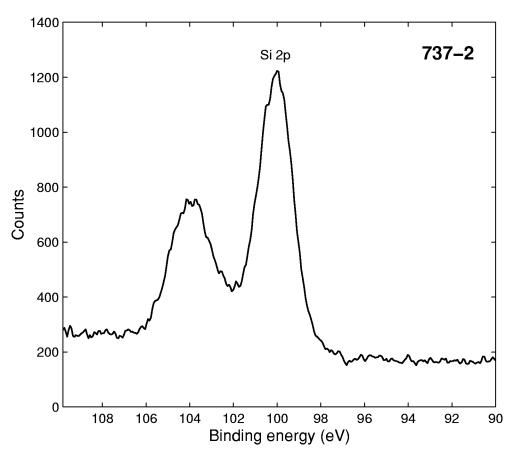
^{*} Voltage shift of the archived (as-measured) spectrum relative to the printed figure. The figure reflects the recommended energy scale correction due to a calibration correction, sample charging, flood gun, or other phenomenon.

b Spot size 800 μ m, pass energy 150 eV, 1 scan. c Spot size 800 μ m, pass energy 150 eV, 3 scans.

^{** [}NP] signifies not published; digital spectra are archived in SSS database but not reproduced in the printed journal.



Accession #	00737-01			
Host Material	Alkyl monolayer on native oxide terminated silicon derived from $ClSi(CH_3)_2(CH_2)_6CH=CH_2$			
Technique	XPS			
Spectral Region	survey			
Instrument	Surface Science SSX-100			
Excitation Source	Al K_{α} monochromatic			
Source Energy	1486.6 eV			
Source Strength	200 W			
Source Size	$0.8~\mathrm{mm} \times 0.8~\mathrm{mm}$			
Analyzer Type	spherical sector			
Incident Angle	55°			
Emission Angle	55°			
Analyzer Pass Energy	150 eV			
Analyzer Resolution	1.5 eV			
Total Signal Accumulation Time	220 s			
Total Elapsed Time	420 s			
Number of Scans	1			
Effective Detector Width	13.0906 eV			
Analyzer Width	$1500~\mu\mathrm{m} \times 12000~\mu\mathrm{m}$			





■ Host Material: Alkyl monolayer on native oxide terminated silicon derived from

CISi(CH₃)₂(CH₂)₆CH=CH₂

■ Technique: XPS ■ Spectral Region: Si 2p

Instrument: Surface Science SSX-100 Excitation Source: Al K_{α} monochromatic

Source Energy: 1486.6 eV Source Strength: 200 W Source Size: $0.8 \text{ mm} \times 0.8 \text{ mm}$

Incident Angle: 55°

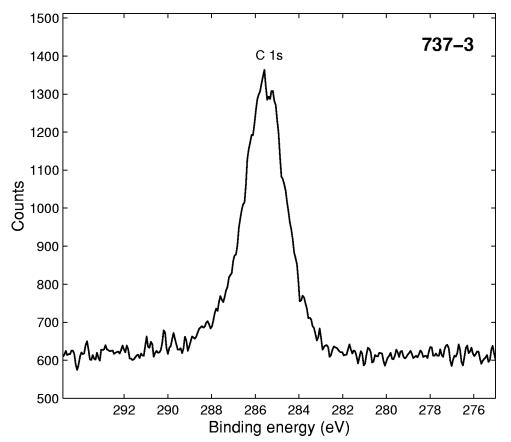
Analyzer Type: spherical sector Analyzer Pass Energy: 150 eV Analyzer Resolution: 1.5 eV

Emission Angle: 55°

Total Signal Accumulation Time: 184 s

Total Elapsed Time: 353 s Number of Scans: 3

Effective Detector Width: 13.0906 eV Analyzer Width: 1500 μ m \times 12000 μ m



■ Accession #: 00737-03

Host Material: Alkyl monolayer on native oxide terminated silicon derived from CISi(CH₃)₂(CH₂)₆CH=CH₂

■ Technique: XPS

■ Spectral Region: C1s

Instrument: Surface Science SSX-100 Excitation Source: Al K_{α} monochromatic

Source Energy: 1486.6 eV Source Strength: 200 W Source Size: $0.8 \text{ mm} \times 0.8 \text{ mm}$

Incident Angle: 55°

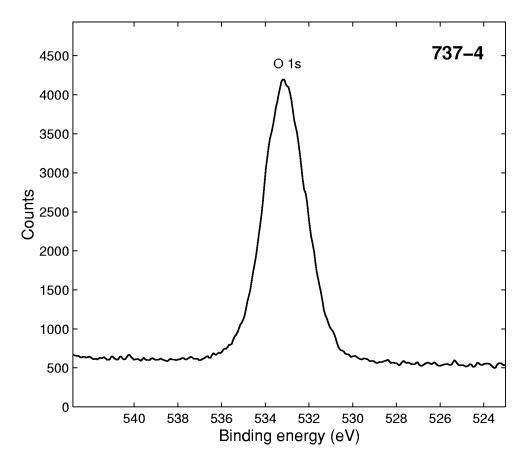
Analyzer Type: spherical sector Analyzer Pass Energy: 150 eV Analyzer Resolution: 1.5 eV

Emission Angle: 55°

Total Signal Accumulation Time: 184 s

Total Elapsed Time: 353 s Number of Scans: 3

Effective Detector Width: 13.0906 eV Analyzer Width: 1500 μ m imes 12000 μ m





■ Host Material: Alkyl monolayer on native oxide terminated silicon derived from CISi(CH₃)₂(CH₂)₆CH=CH₂

■ Technique: XPS■ Spectral Region: O1s

Instrument: Surface Science SSX-100

Excitation Source: Al K_{α} monochromatic

Source Energy: 1486.6 eV Source Strength: 200 W Source Size: $0.8 \text{ mm} \times 0.8 \text{ mm}$

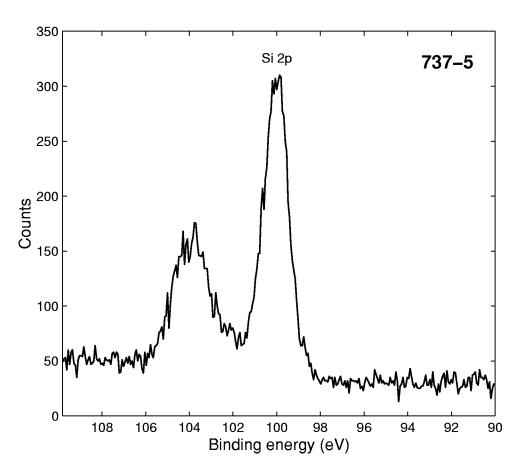
Incident Angle: 55°

Analyzer Type: spherical sector Analyzer Pass Energy: 150 eV Analyzer Resolution: 1.5 eV Emission Angle: 55°

Total Signal Accumulation Time: 184 s

Total Elapsed Time: 353 s Number of Scans: 3

Effective Detector Width: 13.0906 eV Analyzer Width: 1500 μ m imes 12000 μ m



■ Accession #: 00737-05

■ Host Material: Alkyl monolayer on native oxide terminated silicon derived from CISi(CH₃)₂(CH₂)₆CH=CH₂

■ Technique: XPS■ Spectral Region: Si 2p

Instrument: Surface Science SSX-100

Excitation Source: Al K_{α} monochromatic

Source Energy: 1486.6 eV Source Strength: 200 W Source Size: $0.8~\text{mm} \times 0.8~\text{mm}$

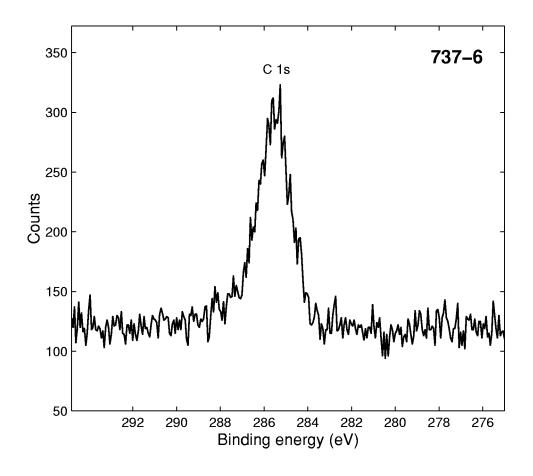
Incident Angle: 55°

Analyzer Type: spherical sector
Analyzer Pass Energy: 50 eV
Analyzer Resolution: 0.5 eV
Emission Angle: 55°

Total Signal Accumulation Time: 184 s

Total Elapsed Time: 353 s Number of Scans: 3

Effective Detector Width: 5.442 eV Analyzer Width: 750 μ m \times 6000 μ m



■ Accession #: 00737-06

■ Host Material: Alkyl monolayer/Si – $CISi(CH_3)_2(CH_2)_6CH=CH_2$

■ Technique: XPS ■ Spectral Region: C1s

Instrument: Surface Science SSX-100

Excitation Source: Al Ka monochromatic

Source Energy: 1486.6 eV Source Strength: 200 W Source Size: $0.8 \text{ mm} \times 0.8 \text{ mm}$

Incident Angle: 55°

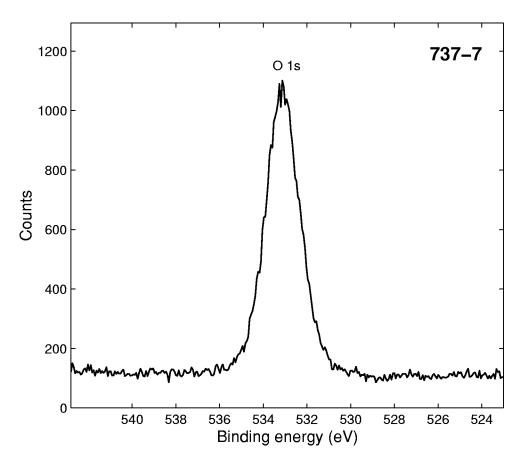
Analyzer Type: spherical sector Analyzer Pass Energy: 50 eV Analyzer Resolution: 0.5 eV

Emission Angle: 55°

Total Signal Accumulation Time: 184 s

Total Elapsed Time: 353 s Number of Scans: 3

Effective Detector Width: 5.442 eV Analyzer Width: 750 μ m imes 6000 μ m



■ Accession #: 00737-07

■ Host Material: Alkyl monolayer/Si – $CISi(CH_3)_2(CH_2)_6CH=CH_2$

■ Technique: XPS ■ Spectral Region: O1s

Instrument: Surface Science SSX-100

Excitation Source: Al Ka monochromatic

Source Energy: 1486.6 eV Source Strength: 200 W Source Size: $0.8 \text{ mm} \times 0.8 \text{ mm}$

Incident Angle: 55°

Analyzer Type: spherical sector Analyzer Pass Energy: 50 eV Analyzer Resolution: 0.5 eV

Emission Angle: 55°

Total Signal Accumulation Time: 184 s

Total Elapsed Time: 353 s Number of Scans: 3

Effective Detector Width: 5.442 eV Analyzer Width: 750 μ m imes 6000 μ m